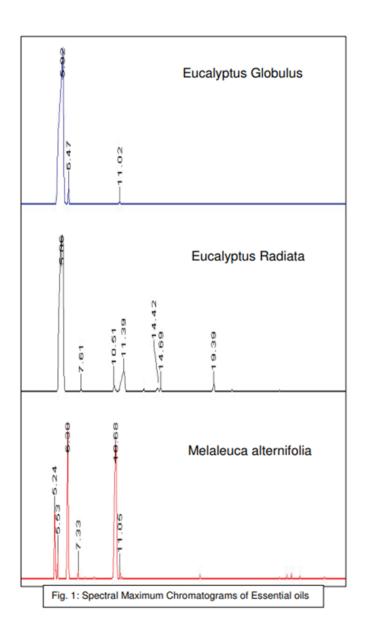


APPLICATION NOTE 004

Essential Oils Analyzed by GC-IR



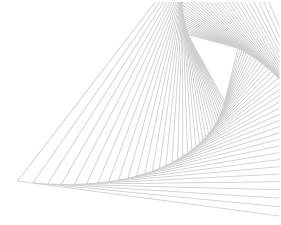
Essential oils of plants from every corner the world are the foundation for many fragrance, flavoring and "nutriceutical" products found in the marketplace. Often the chemicals which give an oil its distinctive qualities are low in abundance compared to the most prevalent species, and can be very similar in structure to components known from related oils previously analyzed. Analysis of these natural products can challenge the state of the art in the analytical lab. Infrared spectral analysis can provide a complementary, "orthogonal" detection method that can lend greater confidence to the identification process. This brief note demonstrates the ability to discriminate among structural isomers by IR spectra alone

For this series of analyses, three Essential Oils [Eucalyptus globulus, Eucalyptus radiata, & Melaleuca alternifolia (tea tree oil)] were analyzed by deposition and IR detection after separation GC on a standard DB-5 column. Each exhibits a distinctive pattern as a Spectral Maximum Chromatogram (SMC) that plots peaks of highest IR absorbance across all wavelengths.

The DiscovIR–GCTM system automatically generates full IR spectra for each peak.

Comparison of the spectra to the solid-phase IR libraries by the DiscovIRTM software yields proposed chemical ID's that correspond to the data from GCMS analysis.



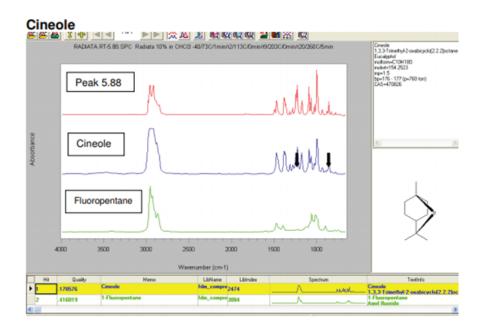


Four commercially available retail fragrances were analyzed by deposition and IR detection after separation GC on a standard DB-5 column. Each exhibits a distinctive pattern as a Spectral Peak Chromatogram (SpC) that plots peaks of highest IR absorbance across all wavelengths. The DiscovIR–GCTM system automatically generates full IR spectra for each peak.

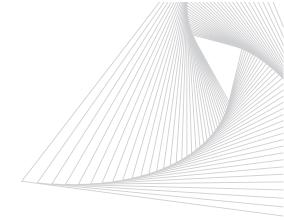
Comparison of the spectra to the solid-phase IR libraries by the DiscovIRTM software can uniquely identify the unknowns. When the potent combination of Mass and IR library matching is used, structure assignments can be made with higher confidence. A key feature of Infrared analysis is that absorption at a certain wavelength is in direct proportion to the amount at hand. Therefore, relative quantities can be determined with confidence, avoiding errors due to disparity in ionization or presence of chromophores.

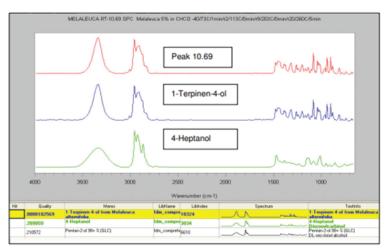
Compound Search using DiscovIR's Infrared library

Using the peak from E. radiata which elutes a 5.88 minutes (and corresponds to peaks in E. globulus & M. alternifolia), the library shows an overall good pattern match with Cineole.





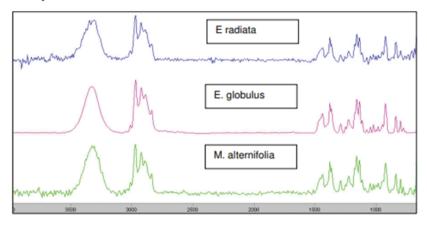




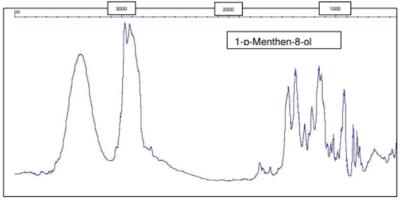
The 10.69 peak from M. Alternifolia was not found in the standard library that comes with the DiscovIRTM system. Research into additional IR libraries along with MS data gave a strong ID of 1-Terpinen-4-ol.

The spectrum was easily added to the DiscovIRTM library so future analyses would display this result.

1-Terpinen-4-ol

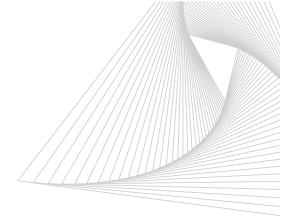


The three spectra here come from corresponding peaks from the three oils. E. radiata 11.39 The peaks from E. globulus at 11.02 min & M. alternifolia 11.05 are relatively small peaks (see figure 1), but the spectra here are of equal quality to the more substantial peak of E. radiata.



This spectrum of 1-p-menthen-8- ol from the literature corresponds closely to those above. Mass data from GCMS confirms the assignment





The data above are from the real-time deposition and detection procedure, even though the peaks are minor species, demonstrating the sensitivity of the system. An added feature of The DiscovIRTM System is the capability to re-scan the deposited track as many times as necessary to reinforce the signal and build a useful spectrum from out of the random noise, providing even better sensitivity when needed.

10% in CHCl Concentration

10% in CHCl₃

1 µL using split/splitless injector, split 50:1

DB5, 20-meter x 0.32-mm ID, 0.25-µm film

Helium carrier, 1 mL/min

70°C for 1 min, 2°/min up to 113°, 9°/min up to 203°, 20°/min to 260°

-40°C